

PRACTICE EXAM 2  
 ITEM 1411 (Chap. 1, 2, 3, 4, 5)

1. Which of the following is a physical change?  
 (a) rusting a nail (b) fruits ripen (c) baking a cake  
 (d) ice melts (e) milk turns sour
2. Which of the following is a homogeneous mixture?  
 (a) water and sand (b) sand, alcohol, and sugar  
 (c) oil and vinegar (d) milk of magnesium (e) air
3. Which of the following is a correct?  
 (a) sodium, S (b) mercury, Mg (c) tin, N, Sn  
 (d) potassium, P (e) gold, Au
4. What is the equivalent temperature for 105 Fahrenheit in Kelvin?  
 (a) 41 (b) 314 (c) 378 (d) 457 (e) -32
5. A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?  
 (a) 1.59 (b) 1.74 (c) 2.36 (d) 2.89 (e) 3.50  
 $d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$
6. How many significant figures are there in the measurement of  $1.0356 \times 10^4$ ?  
 (a) 1 (b) 2 (c) 3 (d) 4 (e) 5
7. How many significant figures are appropriate to show in the result after carrying out the operation below?  
 $(13.7 + 0.027) + 8.221 = ?$   $\times, \div$  least s.f.  $+, -$  least decimal
- (a) 1 (b) 2 (c) 3 (d) 4 (e) 5
8. How many protons, electrons, and neutrons are there in  $^{130}\text{Ba}^{2+}$ ?  
 (a) 56p, 56e, 130n (b) 56p, 54e, 130n (c) 56p, 58e, 74n  
 (d) 56p, 54e, 74n (e) 54p, 56e, 74n
9. Which of the following pairs would have similar chemical and physical properties?  
 (a) Na, Mg (b) H, Li (c) He, Ne (d) Cr, Ca (e) Cl, O
10. Which of the following is the molecular formula?  
 (a) NaCl (b)  $\text{C}_2\text{F}_4$  (c) HCl (d)  $\text{AlBr}_3$  (e) NaCN
11. Which of the following is the empirical formula?  
 (a)  $\text{C}_2\text{H}_4$  (b)  $\text{CH}_2\text{COOH}$  (c)  $\text{P}_2\text{O}_{10}$  (d)  $\text{NH}_4\text{NO}_3$  (e)  $\text{Ag}_2\text{CO}_3$
12. Which of the following is a correct match?  
 (a)  $\text{BaCl}_2$ , barium dichloride (b)  $\text{CS}_2$ , carbon sulfide  
 (c)  $\text{CuF}_2$ , copper (I) fluoride (d)  $\text{HNO}_2$ , nitrous acid  
 (e)  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , magnesium sulfate heptahydrate
13. Which of the following pairs is a correct match?  
 (a)  $\text{Ca}_3(\text{PO}_4)_2$ , calcium (II) phosphate (c)  $\text{Mg}_2\text{O}_3$ , magnesium oxide  
 (b)  $(\text{NH}_4)_2\text{SO}_4$ , ammonium sulfate (d)  $\text{H}_2\text{CO}_3$ , carbon acid  
 (e)  $\text{Cu}(\text{NO}_3)_2$ , copper dinitrate

# Chem 1411 Lab Manual Answer Key

**Emilie Sanchez**



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