

Name: \_\_\_\_\_

For: \_\_\_\_\_

## Section 6.2 Covalent Bonding

(pages 165–168)

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

### Reading Strategy (page 165)

**Relating Text and Models** As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

### Covalent Bonds (pages 165–167)

- Describe a covalent bond.
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
  - electron dot
  - structural formula
  - space-filling
  - electron cloud
- Describe a molecule.
- Is the following sentence true or false?  
 "In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus."
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - one
  - two
  - three
  - four

# Chapter 6 Section 6 2 Covalent Bonding Answer Key

**Paul B. Kelter**



## Chapter 6 Section 6 2 Covalent Bonding Answer Key:

**Chemistry** John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant *Holt Chemistry* Ralph Thomas Myers, 2004

**Principles of Cell Biology** George Plopper, 2014-10-21 Written for undergraduate cell biology courses Principles of Cell Biology Second Edition provides students with the formula for understanding the fundamental concepts of cell biology This practical text focuses on the underlying principles that illustrate both how cells function as well as how we study them It identifies 10 specific principles of cell biology and devotes a separate chapter to illustrate each The result is a shift away from the traditional focus on technical details and towards a more integrative view of cellular activity that is flexible and can be tailored to suit students with a broad range of backgrounds

**Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Chemical Interactions*, 2005 *Applied Chemistry for Polytechnic and Engineering Courses* Dr. R.S. Chauhan, 2021-05-27 The book includes the following chapters in details Language of Chemistry Atomic Structure The Periodic table and Atomic properties Water Chemical Bonding Solutions Electrolysis Environmental Chemistry Experiments

**Quantitative Chemical Analysis** Daniel C. Harris, 2010-04-30 QCA is the bestselling textbook of choice for analytical chemistry It offers a modern portrait of the techniques of chemical analysis backed by a wealth of real world applications This edition features new coverage of spectroscopy and statistics new pedagogy and enhanced lecturer support

**Principles Of Descriptive Inorganic Chemistry** Gary Wulfsberg, 1991-05-29 This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts Translated into Italian *Introduction to Modern Chemistry: Student Manual* Eugene Meyer, 1979

**Biology** Sylvia S. Mader, 2004 *Chemistry, Student Study Guide* John A. Olmsted, Gregory M. Williams, 2005-02-02 100% Pure Chemical Understanding Every morning many of us are energized by a cup of coffee Imagine if you were as

energized by understanding the chemistry in your morning cup from the coffee trees which fill red coffee berries with caffeine and a variety of other chemical substances to the feathery crystals formed by the caffeine molecules to the decaffeinating machines which use liquid solvents to remove this stimulant from some of the beans Now that's real chemical understanding Olmsted and Williams Fourth Edition of Chemistry focuses on helping you see and think about the world and even your coffee as a chemist This text helps you understand how chemical phenomena are governed by what happens at the molecular level apply critical thinking skills to chemical concepts and problems and master the basic mathematical techniques needed for quantitative reasoning You'll see the world as chemists do and learn to appreciate the chemical processes all around us A Fourth Edition with a lot of new perks Revisions include a new early energy chapter revised coverage of bonding expanded coverage of intermolecular forces and increased coverage of multiple equilibria including polyprotic acids New pedagogy strengthens students critical thinking and problem solving skills Visual Summaries at the end of each chapter use molecular and diagrammatic visual elements to summarize essential skills concepts equations and terms eGrade Plus provides an integrated suite of teaching and learning resources including a complete online version of the text links between problems and relevant sections in the online text practice quizzes the Visual Tutor Interactive LearningWare problems and lab demos as well as homework management and presentation features for instructors

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*Bonding Theory* Donald J. Royer, 1968      *Basic Concepts of Chemistry* Leo J. Malone, Theodore O.

Dolter,2012-04-13 Basic Concepts of Chemistry was originally written over thirty years ago to address the needs of general chemistry students with little or no background in chemistry Over time the text has evolved beyond purposes solely aimed at a preparatory chemistry course For some preparatory chemistry students a main sequence in general chemistry may follow but for others a semester of organic and biochemistry may follow Other students enroll to simply satisfy a basic science or chemistry requirement The text was written with a level and functionality designed to accommodate the needs of each of these varied groups of students Basic Concepts of Chemistry was designed with a flexibility that allows instructors to emphasize or omit certain clearly delineated sections The mission of the text has evolved in response to the increased diversity of students and the emphasis on outcomes assessment *Tb Chemistry Molecular Sci* Moore,2001-12

*Complete Preparation for the MCAT* Williams & Wilkins Review,1998-04 Here is the most respected test prep book for the Medical College Admission Test you can buy featuring an active learning approach for a better understanding of the exam s content and a better chance for success Unique to this guide are coverage of all recent changes in the MCAT plus a step by step plan for sharpening cognitive skills developing problem solving skills and critical thinking This thorough guide replaces expensive test preparation courses while giving students exactly what they need to get ready for the MCAT **A Complete Preparation for the MCAT** Aftab S. Hassan,James L. Flowers,1992 This guide for MCAT preparation applies the principles of active and problem based learning to an updated review of content and skills with models for enhanced problem solving and critical thinking abilities There are details on setting up a self managed study programme with guidelines for time management and stress management All areas tested on the exam are covered verbal reasoning physical science writing sample biological sciences with practice questions to chart progress *Histological & Histochemical Methods* John Alan Kiernan,1990 Revised and updated edition 1st was 1981 of a textbook on chemical and physical principles of fixation staining and histochemistry For students i all biological subjects using histological techniques as well as researcher and medical laboratory technologists Annotation copyright Book News Inc *Study Guide for Chemistry, Third Edition* [by] Steven S. Zumdahl Paul B. Kelter,1993

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