

## chapter 6 test review

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. A mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together is called a(n)  
a. dipole. c. chemical bond.  
b. Lewis structure. d. London force.
- \_\_\_\_\_ 2. The electrons involved in the formation of a chemical bond are called  
a. dipoles. c. Lewis electrons.  
b. *s* electrons. d. valence electrons.
- \_\_\_\_\_ 3. As independent particles, most atoms are  
a. at relatively high potential energy. c. very stable.  
b. at relatively low potential energy. d. part of a chemical bond.
- \_\_\_\_\_ 4. Atoms naturally move  
a. toward high potential energy. c. toward less stability.  
b. toward low potential energy. d. away from each other.
- \_\_\_\_\_ 5. If two covalently bonded atoms are identical, the bond is  
a. nonpolar covalent. c. dipole covalent.  
b. polar covalent. d. coordinate covalent.
- \_\_\_\_\_ 6. If the atoms that share electrons have an unequal attraction for the electrons, the bond is called  
a. nonpolar. c. ionic.  
b. polar. d. dipolar.
- \_\_\_\_\_ 7. What are shared in a covalent bond?  
a. ions c. electrons  
b. Lewis structures d. dipoles
- \_\_\_\_\_ 8. The greater the electronegativity difference between two bonded atoms, the greater the percentage of \_\_\_\_\_ in the bond.  
a. ionic character c. metallic character  
b. covalent character d. electron sharing
- \_\_\_\_\_ 9. The pair of elements that forms a bond with the least ionic character is  
a. Na and Cl. c. O and Cl.  
b. K and Cl. d. Mg and Cl.
- \_\_\_\_\_ 10. The B—F bond in  $\text{BF}_3$  (electronegativity for B is 2.0; electronegativity for F is 4.0) is  
a. polar covalent. c. nonpolar covalent.  
b. ionic. d. metallic.
- \_\_\_\_\_ 11. In the three molecules,  $\text{O}_2$ , HCl, and  $\text{F}_2$ , what atom would have a partial negative charge?  
a. oxygen c. chlorine  
b. hydrogen d. fluorine
- \_\_\_\_\_ 12. A neutral group of atoms held together by covalent bonds is a  
a. molecular formula. c. polyatomic ion.

# Chapter 6 Chemistry Test

**G. Philip**



## Chapter 6 Chemistry Test:

**Laboratory and Diagnostic Testing in Ambulatory Care E-Book** Martha (Marti) Garrels, Carol S. Oatis, 2018-08-07  
Selected for Doody's Core Titles 2024 with Essential Purchase designation in Ambulatory Learn to master the latest lab testing techniques Ideal for Medical Assisting and Medical Laboratory Technician programs Laboratory and Diagnostic Testing in Ambulatory Care A Guide for Healthcare Professionals 4th Edition covers the procedures and techniques of commonly used and new CLIA waived point of care tests along with some moderately complex tests Clear step by step instructions and nearly 600 full color photographs make it easy for you to learn each test and procedure This edition includes a focus on new technology and the resulting significant advancements in testing Expanded coverage of electrocardiography and spirometry reflect the expanding roles of Medical Assistants and Medical Laboratory Technicians Review questions in the book are accompanied by practice quizzes and videos online to help you master content and hone skills Comprehensive coverage of the most common CLIA waived tests prepares you for laboratory testing in the ambulatory setting A triad organization gives chapters a consistent easy to follow format with 1 fundamental concepts 2 step by step instructions for CLIA waived procedures and 3 advanced concepts help hone your critical thinking and decision making skills Procedure boxes provide step by step instructions and full color photos and illustrations for today's commonly requested CLIA waived lab tests NEW Content coverage of the latest CLIA waived and moderately complex testing including automated CBCs and automated chemistries and the significant technological advancements NEW Chapter quizzes online help you ensure comprehension and prepare for classroom and certification exams UPDATED art program with more than 100 new and updated images that showcase new technology and the lab testing steps and techniques *Industrial and Engineering Chemistry*, 1925 **Industrial & Engineering Chemistry**, 1923 McGraw-Hill Education SAT Subject Test Chemistry, Fifth Edition Thomas A. Evangelist, 2018-11-23 We Will Help You Get Your Best Score With more than 125 years of experience in education McGraw Hill Education is the name you trust to deliver results This MHE guide is the most comprehensive and relevant SAT Subject Test prep tool on the market This edition provides 5 full length practice tests with thorough answer explanations A comprehensive review of all Chemistry concepts essential to success on the SAT Subject Test An extensive overview of the format of the test based on the most recent SAT Chemistry exams Unique test taking strategies and tips recommended by teachers to help you raise your score A customizable study plan to help you maximize the time you have to prepare TOP 40 LIST The book includes a description of the 40 topics that are most crucial to know before you take the Subject Test in Chemistry TEST TAKING STRATEGIES Learn unique tips developed by teachers to help you avoid the test maker's traps **Essentials Of Medical Laboratory Practice** Constance L Lieseke, Elizabeth A Zeibig, 2012-01-31 Inside you'll find a wealth of information on important laboratory terminology and the procedures you'll need to perform to become an effective member of a physician's office team Coverage of the advanced procedures performed

outside of the physician's office explains what happens to the samples you send out There's also information on CLIA and other government regulations and how they affect each procedure

**Laboratory Testing for Ambulatory Settings - E-Book** Martha (Marti) Garrels, 2014-06-16 Learn the lab testing skills you need to know Laboratory Testing for Ambulatory Settings A Guide for Health Care Professionals 2nd Edition provides in depth coverage of the most common procedures and techniques of all the new CLIA waived point of care tests along with some moderately complex tests Clear step by step instructions and full color photographs make it easy to master each test and procedure Written by noted educators Marti Garrels and Carol S Oatis this edition adds a new chapter on toxicology and information on five new procedures A companion Evolve website lets you practice clinical laboratory skills Complete coverage includes the most common CLIA waived tests for any healthcare professional in the ambulatory setting A triad organization gives chapters a consistent easy to follow format Fundamental Concepts basic information related to tests and procedures CLIA Waived Procedures step by step instructions for CLIA waived tests Advanced Concepts further application of basic knowledge and skills towards a higher level of critical thinking and decision making such as handling non CLIA waived tests Procedure boxes provide clear step by step instructions along with numerous full color photos and illustrations Key terms are defined and reinforced within each chapter Common abbreviations associated with CLIA waived testing are cited at the beginning of chapters 7 10 review questions conclude each chapter to reinforce learning A companion Evolve website includes various activities and exercises to enhance learning with problem solving scenarios A workbook matches the chapters in the textbook offering activities and exercises to reinforce laboratory concepts terminology and procedures Skills sheets help you work through the competency based procedures and meet government standards for good laboratory practice Sold separately Over 60 new photographs and drawings clarify topics and show examples of laboratory specimens so you will be able to identify them on the job New chapter on toxicology More extensive coverage of working with the microscope prepares you for the lab Updated expanded information about quality control and quality assurance provides relevant information so you can accurately and effectively perform in the lab Updates on proper collection and processing of urine specimens microbiology specimens blood capillary and venipuncture specimens which includes new urine culture using vacutainer system new drawings for urine and influenza specimen collections and the latest order of blood draw including the plasma separator tube ensure that you have the most current information Five new procedures with corresponding skill check off sheets help you understand the most up to date protocols Clinitek Analyzer Standard Hematocrit INRatio New A1c Ki iFOB method for fecal occult blood

Phlebotomy Exam Review, Enhanced Edition Ruth E. McCall, 2020-06-19 Phlebotomy Exam Review Enhanced Seventh Edition thoroughly prepares students for any of the national certification exams in phlebotomy

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Develop the strategies that will give you the edge on test day  
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Philip George and son, ltd,1886

**British Medical Journal** ,1930

**AP Chemistry** Albert S. Tarendash, Frederick J. Rowe,2002 This indispensable guide to chemistry helps students who wish to prepare for the AP Chemistry exam on their own Comprehensive and easy to understand this learning guide includes a full content review two full length practice tests with hundreds of practice questions and thorough answer explanations and proven test taking strategies

**The Annual Catalogue of Purdue University, Lafayette, Indiana ... with Announcements for ...** Purdue University,1920

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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Chapter 6 Chemistry Test**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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