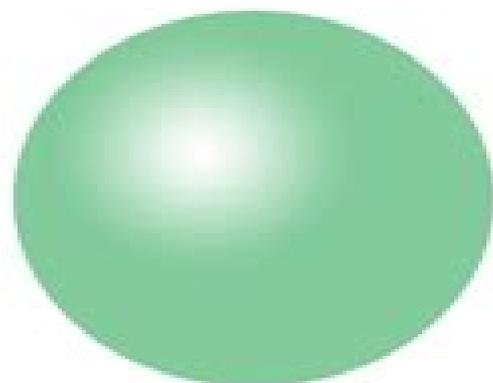


Atomic Radius

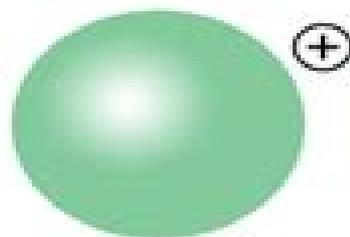
vs

Ionic Radius

Cation: Atomic Radius > Ionic Radius



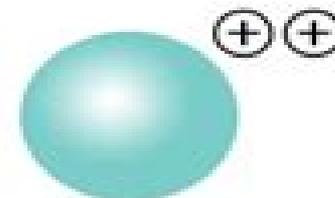
Na  
190 pm



Na<sup>+</sup>  
116 pm



Mg  
145 pm

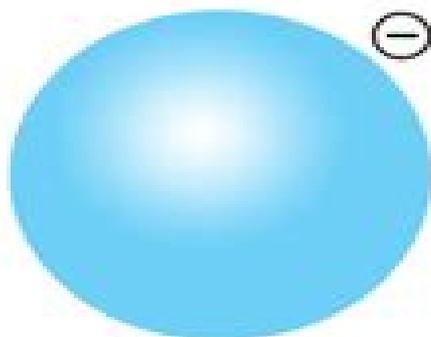


Mg<sup>2+</sup>  
86 pm

Anion: Atomic Radius < Ionic Radius



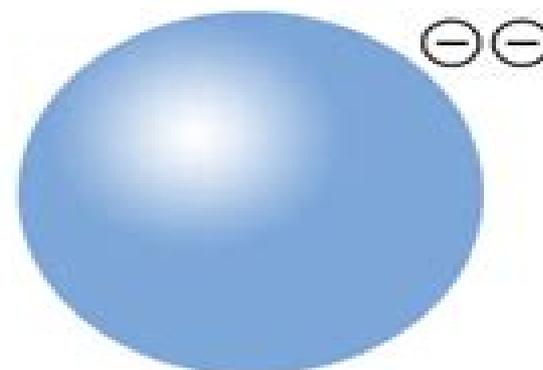
Cl  
79 pm



Cl<sup>-</sup>  
167 pm



S  
88 pm



S<sup>2-</sup>  
184 pm

# Atomic And Ionic Radii Answer Key

**H Kauffman**



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