

IUPAC NOMENCLATURE

Some Common Names



Phenol



Benzaldehyde



Benzoic Acid



Furan



Thiophene



Pyridine



Aniline



Pyridine



Anisole

IUPAC NOMENCLATURE

i° prefix + i° prefix + word root + suffix(s) = suffix (E)
 Secondary Primary Primary Secondary
 cyclic cyclic stem C-C functional
 rings stem C-C group
 Spiro stem C-C

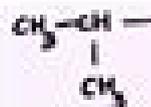
Word Root

1 → meth 3 → prop 5 → pent 7 → hept 9 → non
 2 → eth 4 → but 6 → hex 8 → oct 10 → dec

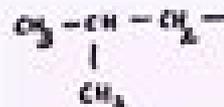
i° prefix → Side Chain | Substituent

-F Fluor
 -Cl Chloro
 -Br Bromo
 -I Iodo
 -NO₂ Nitro

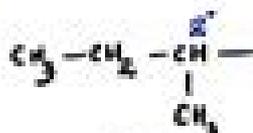
-R (alkyl)
 -CH₃ methyl
 -C₂H₅ ethyl
 -C₃H₇ propyl



isopropyl



isobutyl



sec-butyl



tert-butyl



Neopentyl



vinyl



allyl



cyclopentyl

Rules

1) Select longest continuous carbon chain
 If number of carbon atom in main chain is same, then select that chain which has more branches.

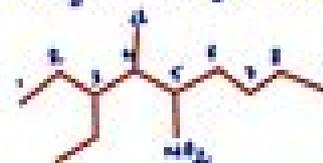
2) Numbering is done in such a way that substituent gets lowest number

If first point of difference is same, then check next point of difference. If substituents getting same number, the alphabetical order is preferred

2 → di = di, tri, tetra are not
 3 → tri used in alphabetical
 4 → tetra order



6-ethyl-2-methyl octane



4-chloro-3-ethyl-4-nitro octane

Activities To Teach Chemistry Nomenclature

TD Snyder



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