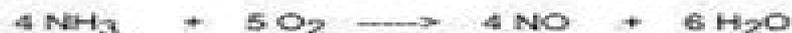


## Stoichiometry Worksheet I

1. Ammonia gas reacts with oxygen gas according to the following equation:



a. How many moles of oxygen gas are needed to react with 23 moles of ammonia?  
(29 mole)

b. How many grams of NO are produced when 25 moles of oxygen gas react with an excess of ammonia? (600 g)

c. If 24 grams of water are produced, how many moles of nitrogen monoxide are formed? (0.89 mole)

d. How many grams of oxygen are needed to react with 6.78 grams of ammonia?  
(16.0 g)

# Chapter 12 Stoichiometry Answer Key

**Neil D. Jespersen, Neil Jespersen, PH.  
D.**

## Chapter 12 Stoichiometry Answer Key:

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