

**UNIT 10: PERIODIC TABLE** **CHAPTER 10: PERIODIC TABLE** **PERIODIC TABLE**

**Section 10.1** (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                            |
|---|----------------------------|
| _____ 18. A column in the periodic table                  | a. metals                  |
| _____ 19. A row in the periodic table                     | b. group                   |
| _____ 20. Group 1 elements                                | c. period                  |
| _____ 21. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 22. Group 17 elements                               | e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of platinum.
- \_\_\_\_\_ 24. Most elements besides boron of the elements in the periodic table are metalloids.
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are located on the left side.
- \_\_\_\_\_ 26. Group 1A elements are the alkali metal metals.
- \_\_\_\_\_ 27. Group 1A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 28. Group 1A elements are (a) nonmetallic elements located on representative metals.
- \_\_\_\_\_ 29. Alkali metals have properties of both metals and more reactive metals.

# Chapter 16 Study Guide For Content Mastery Answers Chemistry

**George Tyler Miller**



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Chemistry Steven S. Zumdahl, Susan A. Zumdahl, 2000 Zumdahl's conceptual approach to problem solving early and thorough treatment of reactions and integration of descriptive chemistry and chemical principles throughout are key elements to the success of this bestseller By encouraging students to understand qualitative concepts before considering quantitative problems Chemistry helps students build a solid understanding of topics and avoid common misconceptions The text also includes new Chemical Impact boxes that present everyday applications of chemistry plus over 200 redesigned illustrations with greater clarity and emphasis on molecular level illustrations **Resources in Education**, 1984

**Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Documentation Abstracts**, 1985

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