

Name _____ Date _____ Class _____

CHAPTER

15

STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is a(n) **(1)** solution of oxygen gas dissolved in nitrogen gas. The oxygen is an **(2)** solute and nitrogen is the **(3)** solvent. Because oxygen gas dissolves in a solvent, oxygen gas is an **(4)** soluble substance. A substance that does not dissolve is **(5)** insoluble. **(6)** Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are **(7)** miscible. However, if one liquid is insoluble in another, the liquids are **(8)** immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



9. Hydration is hydration in which the solvent is water. What is hydration?
Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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85

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CHAPTER

15

Section 15.1 continued

16. An sodium chloride dissolves in water. What happens to the sodium and chloride ions? **The sodium and chloride ions are separated and surrounded by the water molecules.**

17. Explain the orientation of the water molecules around the sodium ions and chloride ions. **Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.**

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? **How do you know?**

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

19. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write *true* or *false*.

- true** 14. The overall energy change that occurs when a solution forms is called the heat of solution.
- false** 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.
- false** 16. Solvation continues as long as the solvation rate is less than the crystallization rate.
- true** 17. In a saturated solution, solvation and crystallization are in equilibrium.
- true** 18. Additional water can be dissolved in an unsaturated solution.
- true** 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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86

Chapter 15 Study Guide For Content Mastery Solutions Answers

**William C. Himstreet, Wayne Murlin
Baty, Carol M. Lehman**



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Basic Pharmacology for Nurses 16 Bruce D. Clayton, Michelle Willihnganz, 2013-01-01 Basic Pharmacology for Nurses is a modern classic nursing pharmacology textbook known for its impeccably accurate drug content and its practical applications of the nursing process A consistent emphasis on health promotion through monitoring and patient education is a hallmark of the book Introductory units ground the reader in basic principles of pharmacology and medication administration Subsequent body system units apply the nursing process to every major disorder and appropriate nursing implications are discussed for every drug class to promote safe medication administration Provides an overview of relevant pathophysiology with an application of the nursing process in each drug chapter followed by nursing implications for each applicable drug class Covers the pharmacology of all major classes of drugs emphasizing side effects to expect side effects to report drug interactions and more Emphasizes patient education and health promotion with easy to recognize Patient Teaching boxes and reproducible Patient Self Assessment forms on the Evolve site Provides need to know coverage of herbal therapies and dietary supplements with an Herbal and Dietary Supplement Therapy chapter and Complementary and Alternative Therapies boxes throughout Increased emphasis on medication safety prepares nurses to practice more safely in a clinical environment that is both more pressured and more focused than ever on reducing drug errors The increased emphasis includes Chapter 7 Principles of Medication Administration do not confuse icons and an icon to signal ISMP s high alert drugs As the focus on pharmacology on the NCLEX continues to increase chapter ending Get Ready for the NCLEX R Examination sections include Key Points Additional Learning Resources and Review Questions for the NCLEX Examination Updated coverage of the most recent FDA approvals withdrawals and therapeutic uses provide impeccably accurate and current content to ensure safe drug therapy Updated and streamlined coverage of nursing care with a focus on the need to know material helps students focus on the most current need to know content

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Renegade Kids, Suburban Outlaws Wayne S. Wooden,Randy Blazak,2001 Topics covered include exploring boundary between deviance and criminality in the lives of young people who are deeply involved in the youth culture show how youth culture is not a set of categories so much as it is a dynamic and creative response to the confusions of growing up in modern society Study Guide George Karleskint,1991

Psychology Study Guide Richard B. Stalling,Wayne Weiten,Ronald E. Wasden,2006-02 By Richard Stalling and Ronald Wasden of Bradley University this study guide includes a review of key ideas people and terms complete with fill in the blank short answer and matching questions This is followed by a self quiz made up of multiple choice questions The Study Guide is organized around learning objectives those same learning objectives that are included in the test bank

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Research in Education ,1970-05 Psychology William Buskist,David W. Gerbing,1990 Business Education Forum ,1981

Prentice Hall Chemistry ,2000 **Abstracts of Instructional Materials in Vocational and Technical Education** ,1969

Système-D 4.0 Frank Dominguez,James S. Noblitt,2004-06 The SYSTEME D WRITING ASSISTANT Software program provides learners with rapid access to language reference materials

Best's Insurance News ,1960 Tapping the Government Grapevine Judith Schiek Robinson,1993 Includes government produced print resources database electronic bulletin boards audiovisual materials archival collections government experts clearinghouses and laboratories

Index to Media and Materials for the Mentally Retarded, Specific Learning Disabled, Emotionally Disturbed National Information Center for Special Education Materials,1978

Resources in Education ,1992

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Table of Contents Chapter 15 Study Guide For Content Mastery Solutions Answers

1. Understanding the eBook Chapter 15 Study Guide For Content Mastery Solutions Answers
 - The Rise of Digital Reading Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 15 Study Guide For Content Mastery Solutions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Personalized Recommendations

Chapter 15 Study Guide For Content Mastery Solutions Answers

- Chapter 15 Study Guide For Content Mastery Solutions Answers User Reviews and Ratings
- Chapter 15 Study Guide For Content Mastery Solutions Answers and Bestseller Lists
- 5. Accessing Chapter 15 Study Guide For Content Mastery Solutions Answers Free and Paid eBooks
 - Chapter 15 Study Guide For Content Mastery Solutions Answers Public Domain eBooks
 - Chapter 15 Study Guide For Content Mastery Solutions Answers eBook Subscription Services
 - Chapter 15 Study Guide For Content Mastery Solutions Answers Budget-Friendly Options
- 6. Navigating Chapter 15 Study Guide For Content Mastery Solutions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 15 Study Guide For Content Mastery Solutions Answers Compatibility with Devices
 - Chapter 15 Study Guide For Content Mastery Solutions Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Highlighting and Note-Taking Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Interactive Elements Chapter 15 Study Guide For Content Mastery Solutions Answers
- 8. Staying Engaged with Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 15 Study Guide For Content Mastery Solutions Answers
- 9. Balancing eBooks and Physical Books Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 15 Study Guide For Content Mastery Solutions Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Setting Reading Goals Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 15 Study Guide For Content Mastery Solutions Answers
 - Fact-Checking eBook Content of Chapter 15 Study Guide For Content Mastery Solutions Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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